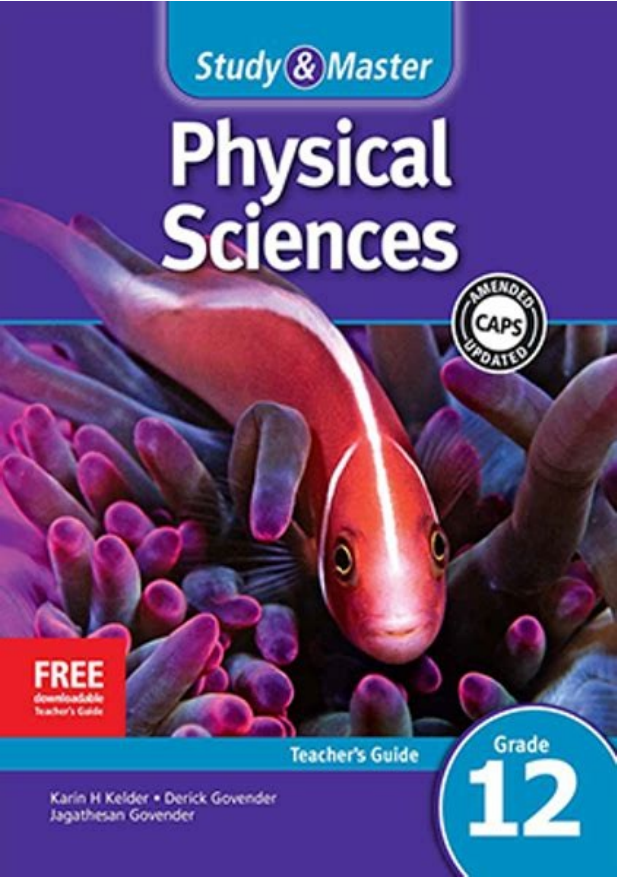
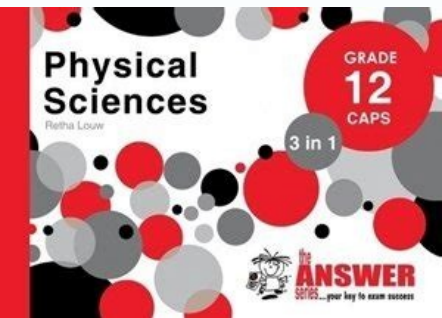
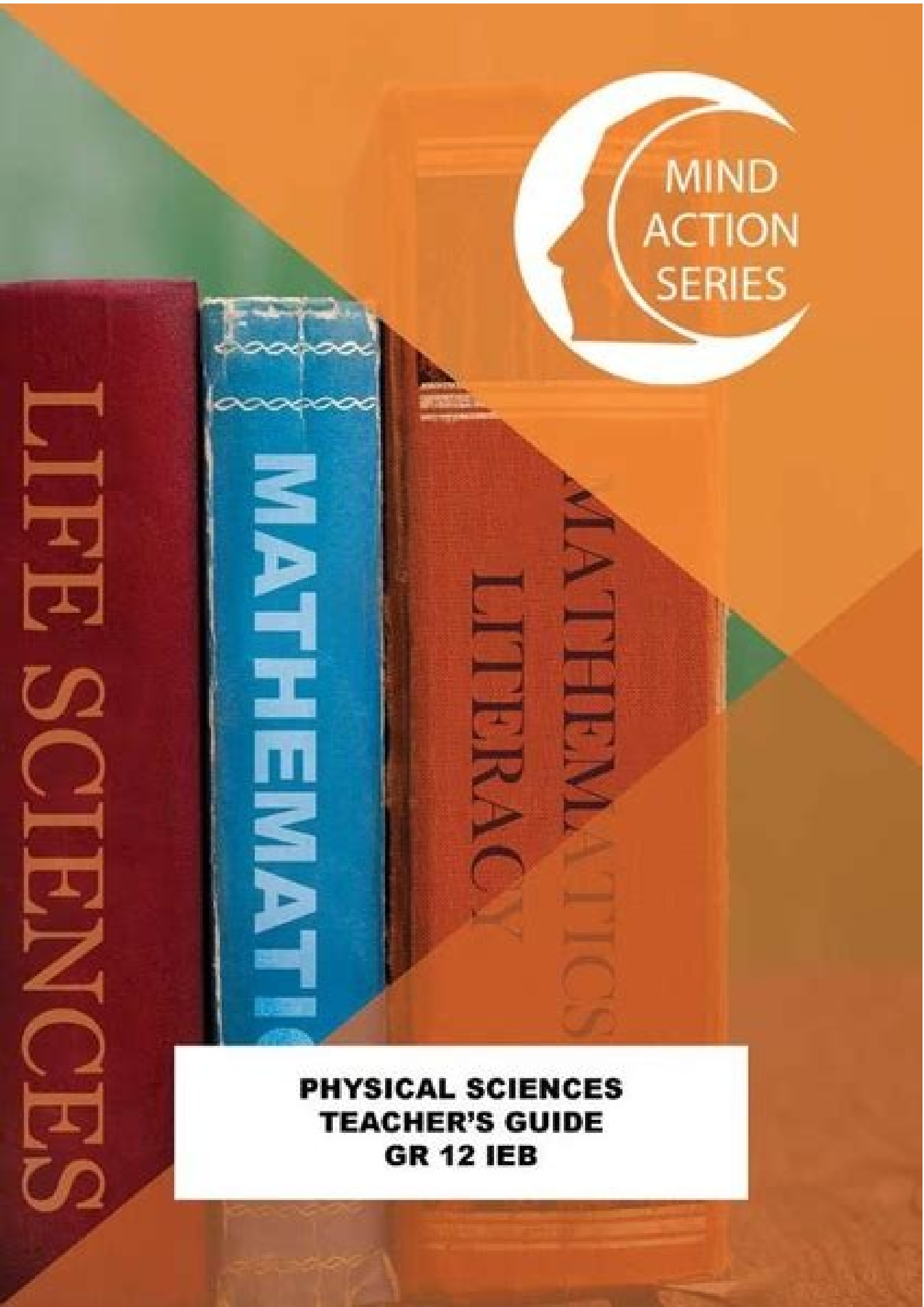
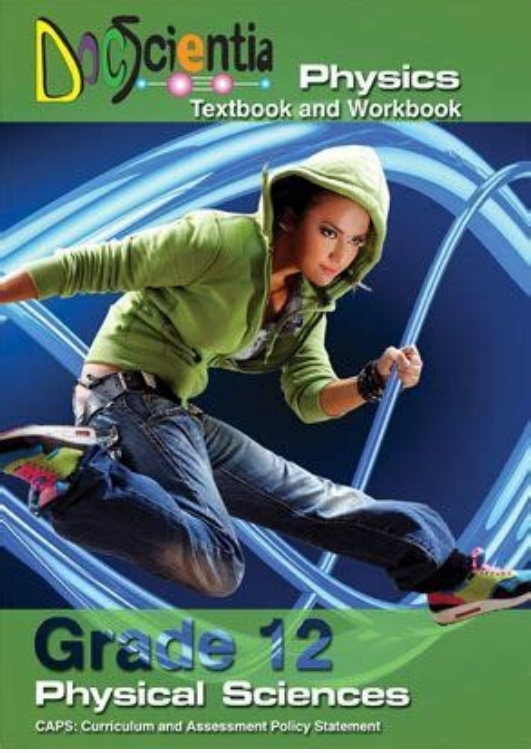


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Velocity vs time graph • Topic 2: Vertical projectile motion in one dimension Fig 2.20 Knowledge area: Mechanics (Physics) a) From the velocity vs time graph, calculate: i) ii) the height of the building. iii) the acceleration of the ball. the distance from the thrower's hand to the highest point. b) Draw the corresponding displacement vs time graph that would have been obtained from the above information. The graph must be to scale. Fill in corresponding values on the x and y axes. c) Calculate: i) the velocity at 1.5 s. ii) the velocity at 2 s. Solution: The graph shows that, upward is being taken as the positive direction. a) i) $A - B: x = \text{area} = \frac{1}{2}bh = \frac{1}{2}(3.5)(34.3) = 60.03 \text{ m}$ ∴ height = 60.03 – 11.03 = 49 m a = gradient $\Delta y = \Delta x (-34.3 - 14.7) = \frac{(5 - 0)}{(5 - 0)} = -9.8 \text{ m.s}^{-2} = 9.8 \text{ m.s}^{-2} \text{ (down)}$ c) i) At 1.5 s, the ball is at its highest point, therefore its velocity is 0 m.s⁻¹. Ay ii) $v = \text{gradient} = \frac{\Delta x (0 - 9.8)}{(4 - 2)} = -4.9 \text{ m.s}^{-1}$ b) -1 = -4.9 m.s⁻¹ down Fig 2.22: Displacement vs time graph Topic 2: Vertical projectile motion in one dimension • 75 Checkpoint 6 A skydiver jumps out of an aeroplane. The first 40 seconds of his motion is plotted on the velocity vs time graph in Figure 2.23. a) Describe the motion of the skydiver from: i) A to B ii) B to C iii) C to D b) Calculate the skydiver's displacement from A to B. c) Calculate the skydiver's acceleration from A to B. d) Calculate the skydiver's displacement from C to D. Fig 2.23: Velocity vs time graph Exercise 2.4 1. A ball is dropped to the ground and when it bounces, it does not bounce all the way to the point from which it was dropped. The path of the ball is shown in the sketch in Figure 2.24a. The graph of velocity vs time for the motion of the ball is given in Figure 2.24b. Fig 2.24a 76 Fig 2.24b: Velocity vs time graph • Topic 2: Vertical projectile motion in one dimension Knowledge area: Mechanics (Physics) a) Without the use of equations of motion, calculate: i) the distance travelled from A to B. ii) the distance travelled from B to C. iii) the distance x on Figure 2.24a. iv) the acceleration of the ball from A to B. b) Sketch the graph of: i) position vs time for the motion of the ball from A to C. Include values on the x and y axis. ii) acceleration vs time for the motion of the ball from A to C. Include values on the x and y axis. 2. A ball is thrown upward from the top of a building. The path of the ball is shown in the sketch in Figure 2.25a. The graph of velocity vs time for the motion of the ball is given in Figure 2.25b. Fig 2.25a Fig 2.25b: Velocity vs time graph a) Without the use of equations of motion, calculate i) the distance travelled by the ball from A to B. ii) the distance travelled by the ball from B to C. iii) the height of the building. iv) the acceleration of the ball from A to B. Topic 2: Vertical projectile motion in one dimension • 77 b) What will the acceleration of the ball be from B to C? c) Sketch the graph of: i) position vs time for the motion of the ball from A to C. Include values on the x and y axis. ii) acceleration vs time for the motion of the ball from A to C. Include values on the x and y axis. 3. A ball is thrown upward and eventually falls back down to the position from which it was thrown. The path of the ball is shown in the sketch in Figure 2.26a. The graph of velocity vs time for the motion of the ball is given in Figure 2.26b. Fig 2.26a Fig 2.26b: Velocity vs time graph a) What will be the value of the acceleration of the ball from A to B? b) Without the use of equations of motion, calculate the value of t1. c) What is the value of t2? d) What is the value of x1? 78 • Topic 2: Vertical projectile motion in one dimension Knowledge area: Mechanics (Physics) 4. A ball is dropped from rest at point A and bounces twice at B and D, as shown in Figure 2.27a. The graph of velocity vs time is shown in Figure 2.27b. Fig 2.27a Fig 2.27b: Velocity vs time graph a) Without the use of equations of motion, calculate: i) ii) the acceleration from A to B. iii) the distance from B to C. iv) the distance from point A to point C. v) the total distance travelled by the ball from A to E. the distance from A to B. b) Draw the corresponding displacement vs time graph for the motion, including values on the x and y axes. 5. The graph in Figure 2.28 shows the motion of a parachutist from the moment she jumps out of the plane until she hits the ground. Fig 2.28 a) Describe her motion from 0 – 30 seconds. Topic 2: Vertical projectile motion in one dimension • 79 b) What happens at 30 seconds that changes the shape of the graph so significantly? c) From the graph, calculate: i) ii) the distance that the parachutist falls from 0 – 3 seconds. iii) the acceleration of the parachutist from 12 – 30 seconds, iv) the distance that the parachutist falls from 12 – 30 seconds, the acceleration of the parachutist from 0 – 3 seconds. d) Why is the velocity of the parachutist so much slower from 50 – 180 s than from 12 – 30 s? Explain. 6. A ball is thrown upward at 25 m.s⁻¹ from the roof of a building (point A). It hits the ground at point C, as shown in Figure 2.29a. The corresponding displacement vs time graph for the motion is given in Figure 2.29b. Fig 2.29a a) Use equation of motion to find: i) ii) the velocity at C Fig 2.29b: Displacement vs time graph the distance from A to B (fill this value in on the graph) b) What is the value of: i) x1 on the graph? ii) x2 on the graph? 80 • Topic 2: Vertical projectile motion in one dimension Knowledge area: Mechanics (Physics) c) Sketch the velocity vs time graph for the motion, including values on the x and y axes. d) Sketch the acceleration vs time graph for the motion, including values on the x and y axes. 7. A ball is thrown upward from A to its highest point B. The path of the ball is shown in the sketch in figure 2.30a. The graph of position vs time for the motion of the ball is given in Figure 2.30b. Fig 2.30a Fig 2.30b: Position vs time graph Without the use of equations of motion, calculate the velocity of the ball at 0.8 s. Extend yourself 1. A group of hikers come to a kran's (a sheer cliff). The first hiker uses a chain ladder to climb from the soft dry river bed below to the top of the 20 m kran's. The hikers below ask him to send some oranges down to them before they climb the ladder. The oranges strike the soft river bed below. Assume that air resistance is negligible. • He drops the first orange (A) from a height of 20 m. • He sends the second orange (B) down to them, throwing it with a velocity of 10 m.s⁻¹ downward. • He throws a third orange (C) upward into the air with velocity of 10 m.s⁻¹. a) Calculate the time taken for orange A to reach the river bed below. (3) w Fig 2.31 Topic 2: Vertical projectile motion in one dimension • 81 b) Calculate the magnitude of the impact velocity of each of the oranges when it reaches the river bed 20 m below him. Do this for: i) Orange A (4) ii) Orange B (3) iii) Orange C (3) c) Draw velocity vs time graphs (on the same set of axes) for the flight of each of the three oranges. Label them graph A, B and C to correspond with each of the oranges A, B and C. (12) d) In which way are the flights of the three oranges similar? (1) e) In which way do the flights of the three oranges differ? (1) One of the hikers is curious about how the depth of the crater formed by a falling orange varies with the impact velocity of the orange. She asks you to design an experiment to answer his question. f) Write a hypothesis for his question. (2) g) Design an experiment to test your hypothesis. (5) 2. John designed and constructed a model rocket. He placed it on top of a high platform to allow his friends to get the best possible view. He recorded the launch on a video tape. Using the video tape, John was able to plot the velocity vs time graph of the motion of the rocket. Refer to Figure 2.32. a) For the first 3 seconds: i) ii) draw and label a free-body diagram of the forces acting on the rocket. (4) describe the motion of the rocket.(4) Fig 2.32 b) After 3 seconds, all the rocket fuel is burnt up and the rocket is now in free-fall. i) What is the gradient of the graph between t = 3 s and t = 7 s? ii) Determine the time at t.(6) 82 c) Sketch a displacement vs time graph for the motion of the rocket. • Topic 2: Vertical projectile motion in one dimension (1) 5 Knowledge area: Mechanics (Physics) 3. Neil and Susan are interested in the bouncing vertical motion of a ball. They set up an experiment to do this. A ball of mass 250 g is dropped from a fixed height of 2 m. Neil times the time taken from the moment the ball leaves Susan's hand until it touches the floor. He uses a stopwatch. They record the height reached after the bounce by taking photographs of the ball bouncing back to maximum height in front of a 2 m ruler. They repeat these measurements three times and record the following results. The times taken to reach the floor are 0.61 s, 0.65 s and 0.64 s respectively. The corresponding maximum heights of the bounces are 1.54 m, 1.60 m and 1.58 m respectively. a) Draw up a table of the results of this experiment and calculate the average values of time taken and maximum height reached. Include these average values in the table. Make sure that your columns have appropriate headings and SI units. (4) b) What is the magnitude of the acceleration of the ball while it falls to the floor? Ignore the effects of air resistance.(1) Fig 2.33 c) Use the values from the table to calculate the magnitude of the ball's velocity when it reaches the floor. (3) d) Use results from the table to calculate the magnitude of the ball's velocity when it leaves the floor. (3) [54] Definitions free fall. the motion of a body in which the only force acting on it is gravity terminal velocity. the constant speed a free falling object eventually reaches when the air resistance prevents further acceleration Topic 2: Vertical projectile motion in one dimension • 83 Summary • An object is in freefall when the only force acting on it is the force of gravity. • Air resistance causes the acceleration of an object falling vertically to decrease, until eventually the acceleration is zero and the object has reached terminal velocity. Terminal velocity is a constant velocity and acceleration is zero. • These equations can be used to solve for unknown values regarding linear motion with constant acceleration: In symbols: $v_f = v_i + a\Delta t$ $v_f^2 = v_i^2 + 2a\Delta x$ $12 a\Delta t^2 \Delta x = v_i\Delta t + \frac{1}{2} a \Delta t^2$ $\Delta x = \frac{v_f + v_i}{2} \Delta t$ $\Delta x = \frac{v_f^2 - v_i^2}{2a}$ Where: v_i = initial velocity (m.s⁻¹) v_f = final velocity (m.s⁻¹) Δx = distance/displacement (m) Δt = time (s) a = acceleration (m.s⁻²) • When doing calculations from a velocity vs time graph: - Area under the graph = displacement or distance - Gradient of the graph = acceleration • When doing calculations from a displacement vs time graph: 84 - Gradient of the graph = velocity • Topic 2: Vertical projectile motion in one dimension Topic 3 Organic chemistry What you will learn about in this topic • Organic molecular structures • Structure and physical property relationships • Applications of organic chemistry • Types of reactions of organic compounds • Plastics and polymers Let's talk about this topic The photo shows an oil drill. Oil is a vital resource in the world today. From it we are able to produce fuels such as petrol, as well as other products such as road tar and plastics. The world simply cannot function without oil and its derivatives. However, it is believed that there is only enough oil reserves left in the world for another 35 to 40 years, if we continue consuming it at the current rate that we are. That is why there is so much interest in alternative energy resources. Oil is an example of an organic compound. You will learn more about organic compounds in this topic. Topic 3: Organic chemistry • 85 Organic molecular structures What you know already In Grade 11, in the topic 'Exploiting the lithosphere', you learnt about the use of coal and oil as fossil fuels. When these fossil fuels burn, a chemical reaction occurs that produces greenhouse gases that result in pollution and global warming. The combustion of fuels is just one type of reaction that we will investigate in this topic. k Chec If myse 1. What gases are formed when coal and oil are burned? 2. Are coal and oil renewable or non-renewable resources? Explain. Introduction to organic molecular structures Organic chemistry can be considered to be the chemistry of carbon compounds. Together with carbon, hydrogen is present in most organic compounds - these substances are called hydrocarbons. Organic compounds can also contain other elements along with carbon, such as oxygen, nitrogen, chlorine and bromine. Carbon is the basic building block of organic compounds. Carbon recycles through the Earth's atmosphere, water, soil and living organisms, including human beings through the carbon cycle. Carbon is present in our bodies in the form of proteins, carbohydrates and fatty acids, whilst carbon is also present in other organic substances such as petrol and plastic. Fig 3.1: Carbon is present in all these items. There are millions of different organic materials and substances on Earth. The reason for this is because of carbon's unique bonding capabilities. Carbon has the unique characteristic among all elements to form long chains of its own atoms, a property called catenation. It is able to form very long chains of carbon atoms, as well as branches and rings, as shown in Figure 3.2: a) A straight-chain hydrocarbon b) A branched-chain hydrocarbon Fig 3.2: Carbon atoms can form chains, branches and rings 86 • Topic 3: Organic chemistry c) A hydrocarbon ring Knowledge area: Matter and materials (Chemistry) Carbon can also form double and triple bonds between carbon atoms, as shown in Figure 3.3. Some factors which make carbon - carbon bonds unique include: • The fact that the covalent bond between two carbon atoms is quite strong. a) A hydrocarbon compound containing a double bond b) A hydrocarbon compound containing a triple bond Fig 3.3: Carbon atoms can form double and triple bonds. • Carbon compounds are not extremely rare under ordinary conditions. • A wide variety of carbon compounds are possible since carbon can form up to four single covalent bonds. • The ability of carbon to make bonds with itself - a process known as catenation. • The ability of carbon to make multiple bonds with itself. Organic compounds can be represented in different ways. You need to know the following methods of representation: • Molecular formula, e.g. C4H10 This is the simplest method of representing an organic substance. It only shows how many atoms of each element there are within the molecule, but it gives us no indication as to where the atoms are found in relation to one another. • Structural formula, e.g. This method shows us exactly where each atom within the molecule is found. • Condensed structural formula, e.g. CH3CH2CH2CH3 This method lists the carbon atoms in the molecule individually and shows how many hydrogen atoms are connected to each carbon atom without showing the bonds. Checkpoint 1 Write down whether the following examples are showing the molecular formula, the condensed structural formula or the structural formula for the organic compound. 1. CH3CH2CH3 2. C3H8 3. Topic 3: Organic chemistry • 87 We will study organic compounds that are grouped together, as they have similar characteristics and properties. It is necessary to classify the millions of organic compounds that exist in some way. We do this by identifying the functional group and homologous series of the compound. The functional group of a compound is an atom or a group of atoms that form the centre of chemical activity in the molecule. A homologous series is a series of similar compounds which have the same functional group and whose consecutive members differ by - CH2 in their molecular formula. All compounds in the same homologous series obey the same general formula. The functional group of a compound assists us in identifying to which homologous series an organic molecule belongs. Science around us Rock paintings Charcoal, which consists mostly of carbon, was used by the San people to mix black paints used in rock paintings, as shown in Figure 3.4. - - - C - C - - Alkanes Fig 3.4: Rock paintings made using charcoal. The alkanes are a homologous series of hydrocarbons where the molecules are characterised by single bonds between their carbon atoms. Alkanes are said to be saturated compounds. A saturated organic compound is one that only contains single bonds attached to carbon atoms that make up the compound. The functional group of all - alkanes is - C - C -, as this indicates that all the carbon atoms within the molecule are - - connected by single bonds. Alkanes have the general formula CnH2n + 2. The general formula can be used to determine how many hydrogen atoms an alkane molecule will contain, if the number of carbon atoms within the molecule is known. For example, if the alkane contains 3 carbon atoms, then n=3. If n = 3, then the number of hydrogen atoms within the molecule will be 2(3) + 2 = 8. Therefore the molecular formula of the compound will be C3H8. The name of the alkane is determined by the number of carbon atoms it contains in the molecule. The number of carbon atoms in the molecule determines the prefix of the name. Alkane molecules' names will always end with the suffix '-ane', indicating that they are alkanes. Number of carbons in main carbon chain and prefix: 1- meth 5- pent 2- eth 6- hex 3- prop 7- hept 4- but 8- oct 88 • Topic 3: Organic chemistry Knowledge area: Matter and materials (Chemistry) Table 3A shows some alkanes. Table 3A Condensed Structural Formula Molecular Formula methane CH4 CH4 ethane CH3CH3 CH2H6 propane CH3CH2CH3 C3H8 Name Structural Formula The structural formula of alkanes do not necessarily have to be drawn in a straight line. For example, octane can also be drawn as shown in Figure 3.5. However, it is easiest to draw the carbon atoms in a straight line when drawing structural formulae. Checkpoint 2 1. Draw the structural formula for: a) butane Fig 3.5: Octane b) hexane. 2. Use the general formula for alkanes to determine the molecular formula for: a) pentane b) octane. Branched alkanes We already know that organic molecules may have a branched-chain structure. We will now study branched alkanes. To name the branches, we will need to indicate in the name the number of carbon atoms in each branch. Table 3B indicates the name of a branch, determined by how many carbon atoms the branch contains. These are known as alkyl groups, hence the name of the branch will always end in -yl. Topic 3: Organic chemistry • 89

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